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Chemistry Chapter 6 Assessment. As you move across the number of protons in the nucleus increase; however the distance between the nucleus and the valence electrons remains the same. Thus the protons have a stronger pull on the valence electrons decreasing the size of the atom.

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molar mass of $C_6H_6O = 94.11 \text{ g}$ 21. a. molar mass of $SO_3 = 80.07 \text{ g}$ 49.2 mg SO_3 1 g 1000 mg 1 mol 80.07 g = 6.14 10^{-4} mol SO_3 b. molar mass of $PbO_2 = 239.2 \text{ g}$ 7.44 x 104 kg PbO_2 2 1000 g 1 kg 1 mol 239.2 g = 3.11 105 mol PbO_2 c. molar mass of $CHCl_3 = 119.37 \text{ g}$ 59.1 g $CHCl_3$ 1 mol 119.37 g = 0.495 mol $CHCl_3$ d. molar mass of $C_2H_3Cl_3 = 133.39 \text{ g}$ 3.27 ...

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The Periodic Table and Periodic LawThe Periodic Table and ...

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Answer to 12. Each period in the periodic table corresponds to . a. an energy level c. an orbital b. an electron d. a suborbital

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